

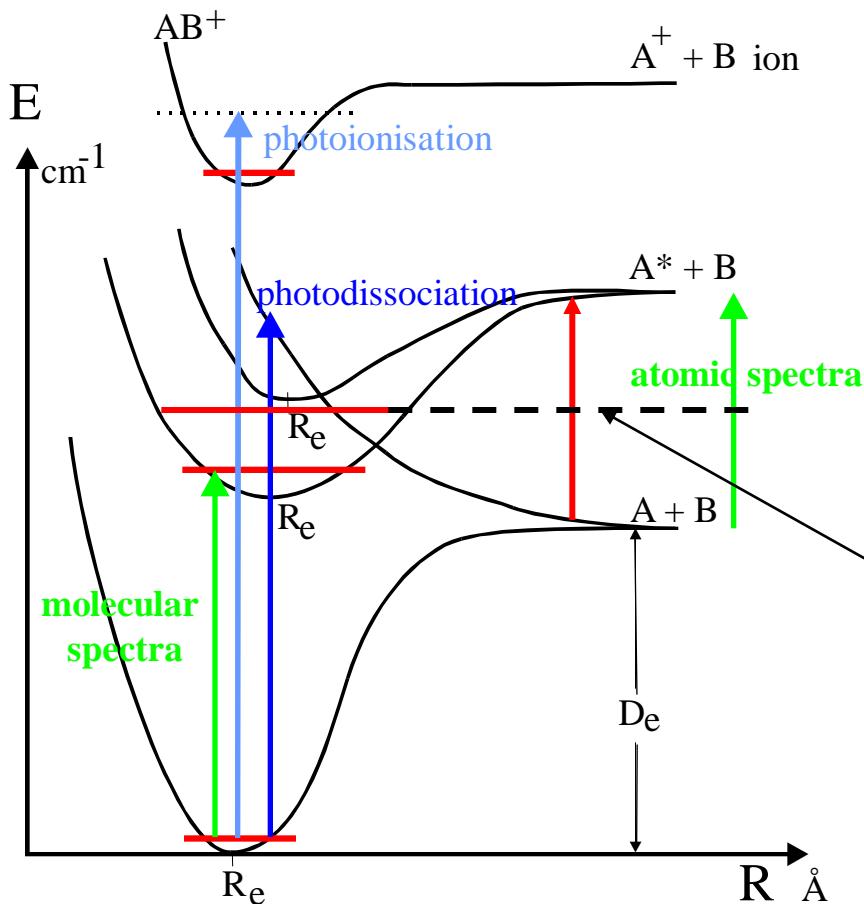
Renaissance in diatomic spectroscopy

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Molecule AB



molecular spectra

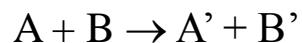
quantum numbers, configuration, potentials

photodissociation, “half” collision

**photoionisation, photoelectrons, ZEKE,
autoionisation**

ionisation energy

potential crossing, coupling of states
predissociation
Feshbach resonance

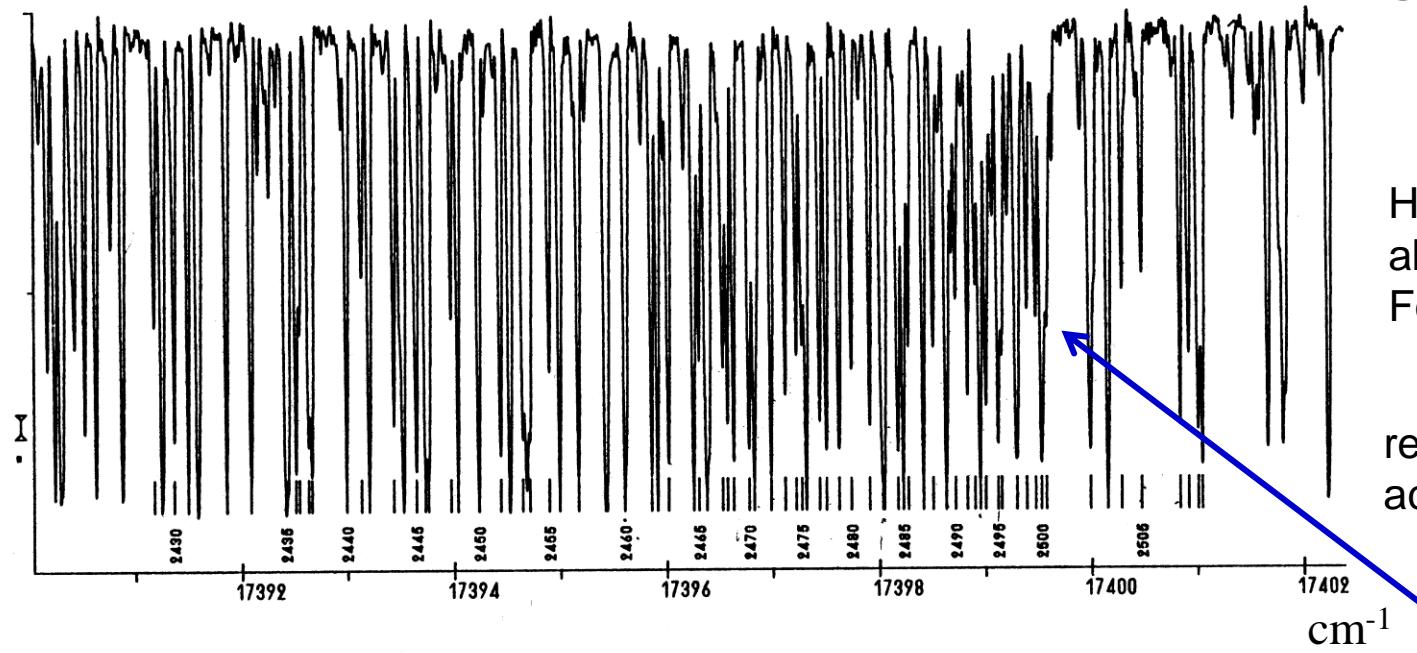


collisions, collisions in fields, optical collisions, etc.

Band spectra and requirements on resolution

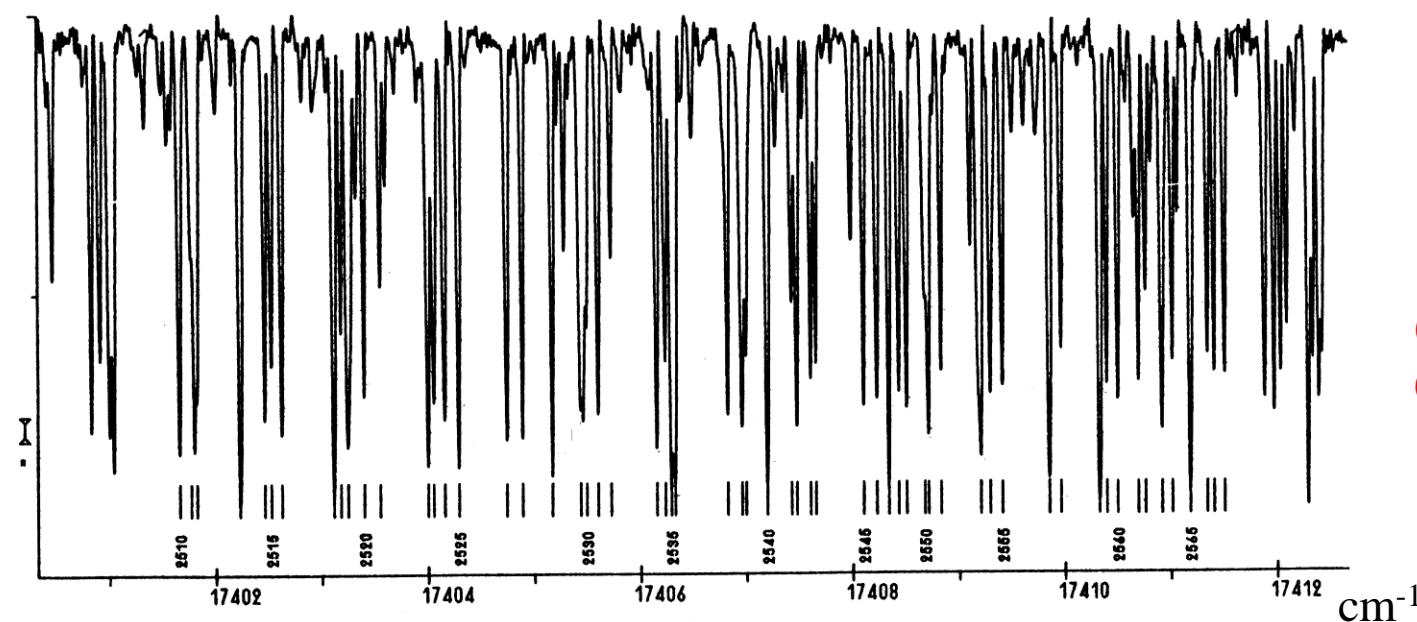
- classical spectroscopy:
 - diffuse bands
 - band heads and vibrational systematic
 - shaded bands and rotational energy
 - molecular parameters
 - Herzberg: $\omega_e, \omega_e x_e, \dots, B_e, \alpha_e, \dots$
 - Dunham: power series in $v+1/2$ and $J(J+1)$ Y_{kl}
 - compact form of large data sets of rovibrational structure
 - microwave spectroscopy:
 - rotational energies, hyperfine structure, Stark- and Zeeman effect
 - laser spectroscopy:
 - high resolution, Doppler reduced, non-linear methods etc.
 - looking with a magnifying glass
 - Fourier transform spectroscopy
 - Ultracold matter as new kind of ensemble
- Varying the application of light*

Some remarks to the state of the art in conventional spectroscopy



resolution 0.01cm^{-1}
 accuracy 0.003cm^{-1}

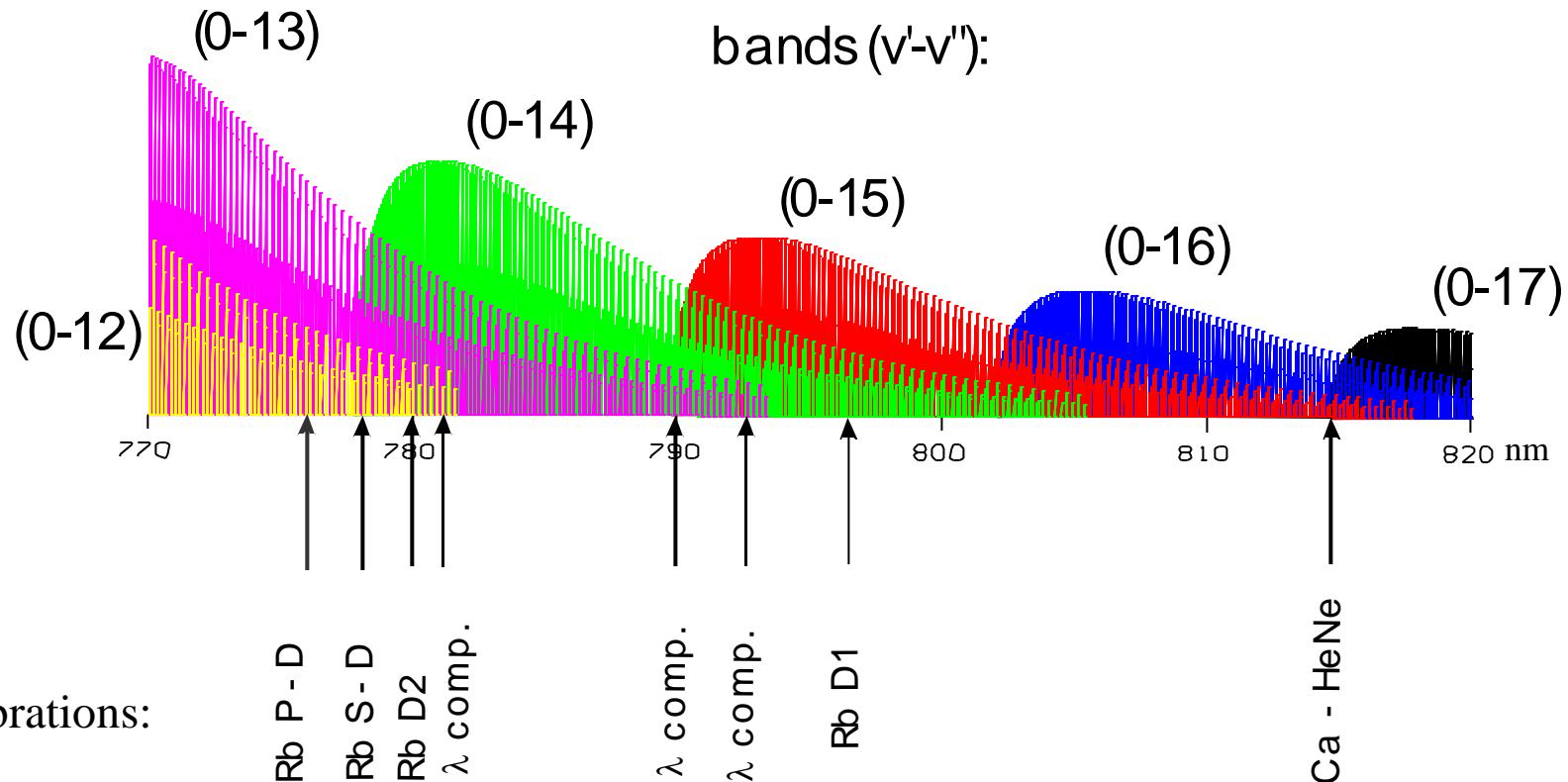
dense!



Iodine atlas
 Gerstenkorn/Luc
 Orsay

Synthetic band spectra

example I₂ T = 900K



Spectra simulation of optical spectrum of Iodine

iodine atlas on the notebook
poor man's secondary frequency standard

accuracy: in general 0.003cm^{-1} or 100MHz
areas better than 10 MHz and some better than 1MHz
 $\Delta\nu/\nu$ up to 10^{-9}

reproducibility: depends on the preparation of the absorption cell
pressure shift, temperature dependence

www.iqo.uni-hannover.de

→ Molecules & Lasers → Secondary Frequency Standard

What is behind such simulation?

huge data set of high precision from several sources

← checked internal consistency!

assignment of spectra

model Hamiltonian:



molecular potentials replace molecular parameters

or

coupled system with potential matrix → “deperturbation”

fitting procedure → compact form of “all” data

by potentials and simple coupling functions

dense ultracold ensemble of atoms

interaction: light and/or collisions

ensemble with atoms and molecules
statistical distribution?

new chances

design the ensemble with reduced dimensionality or spatially ordered

↓
cigar or disk shape

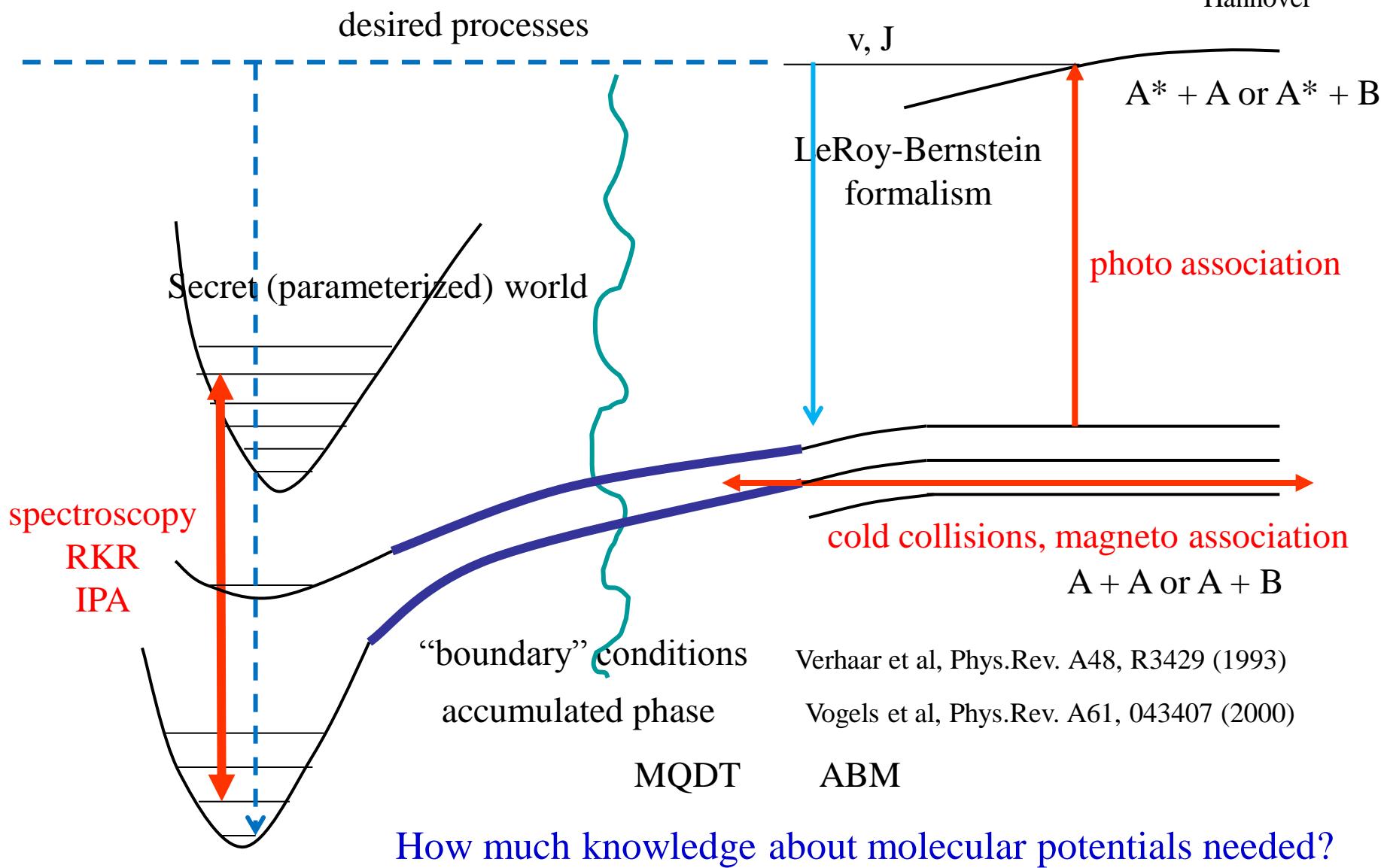
↓
optical lattice

start with pure atomic ensemble
only one or two quantum states

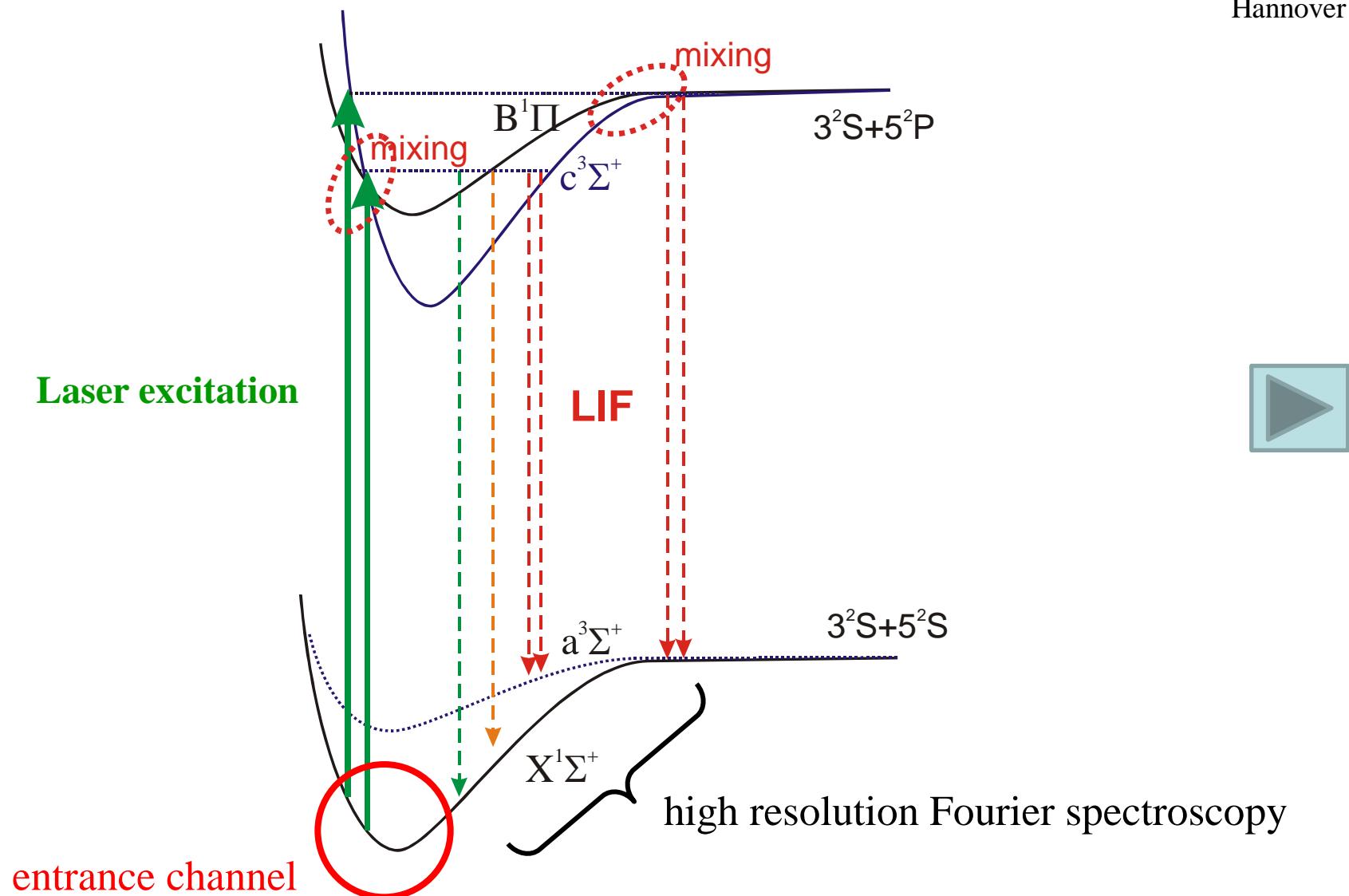
design a quantum specific pathway
to molecules

spectroscopy can and will provide
the data about the interactions

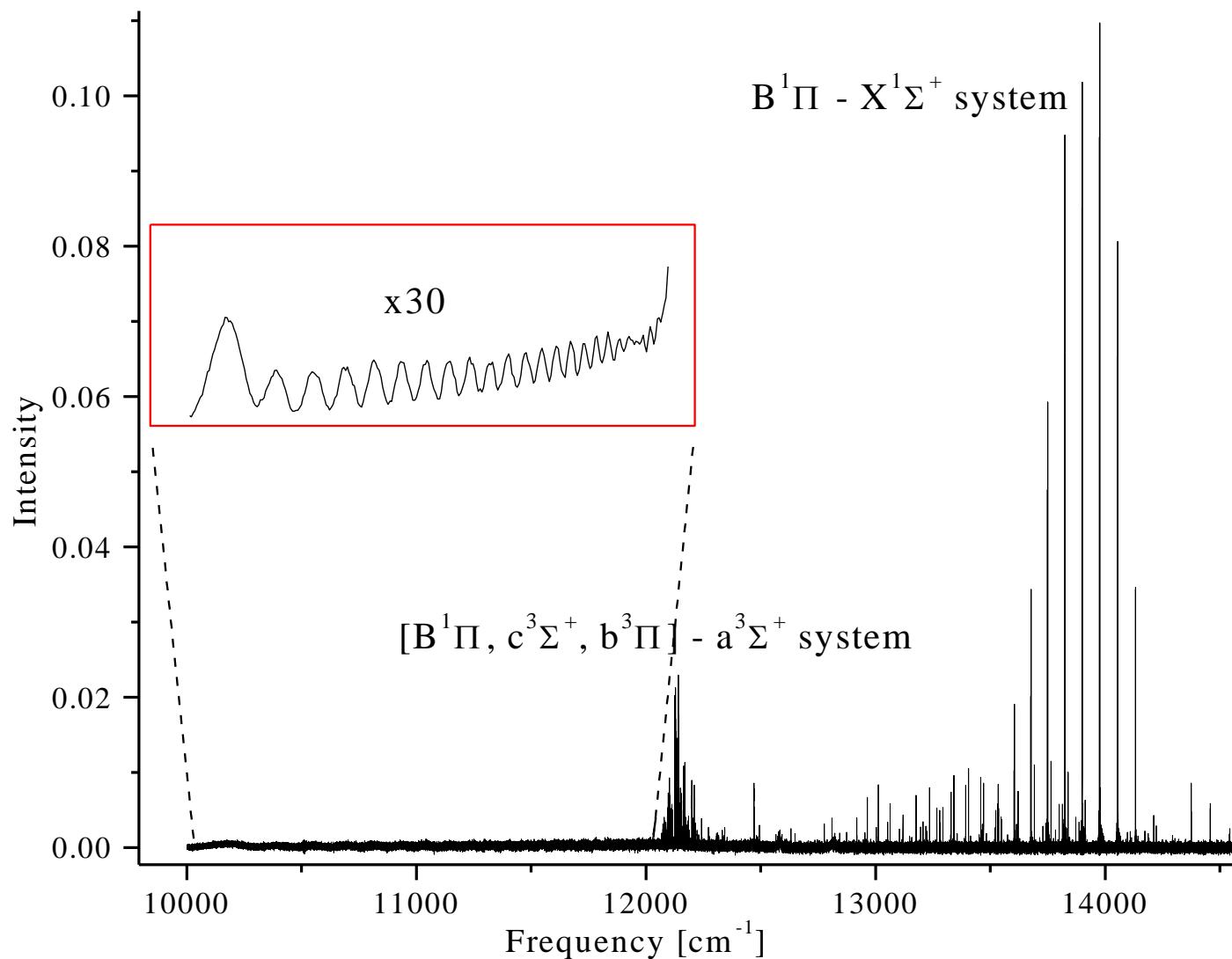
Different experimental situations!



The spectroscopic workhorse



Example NaRb

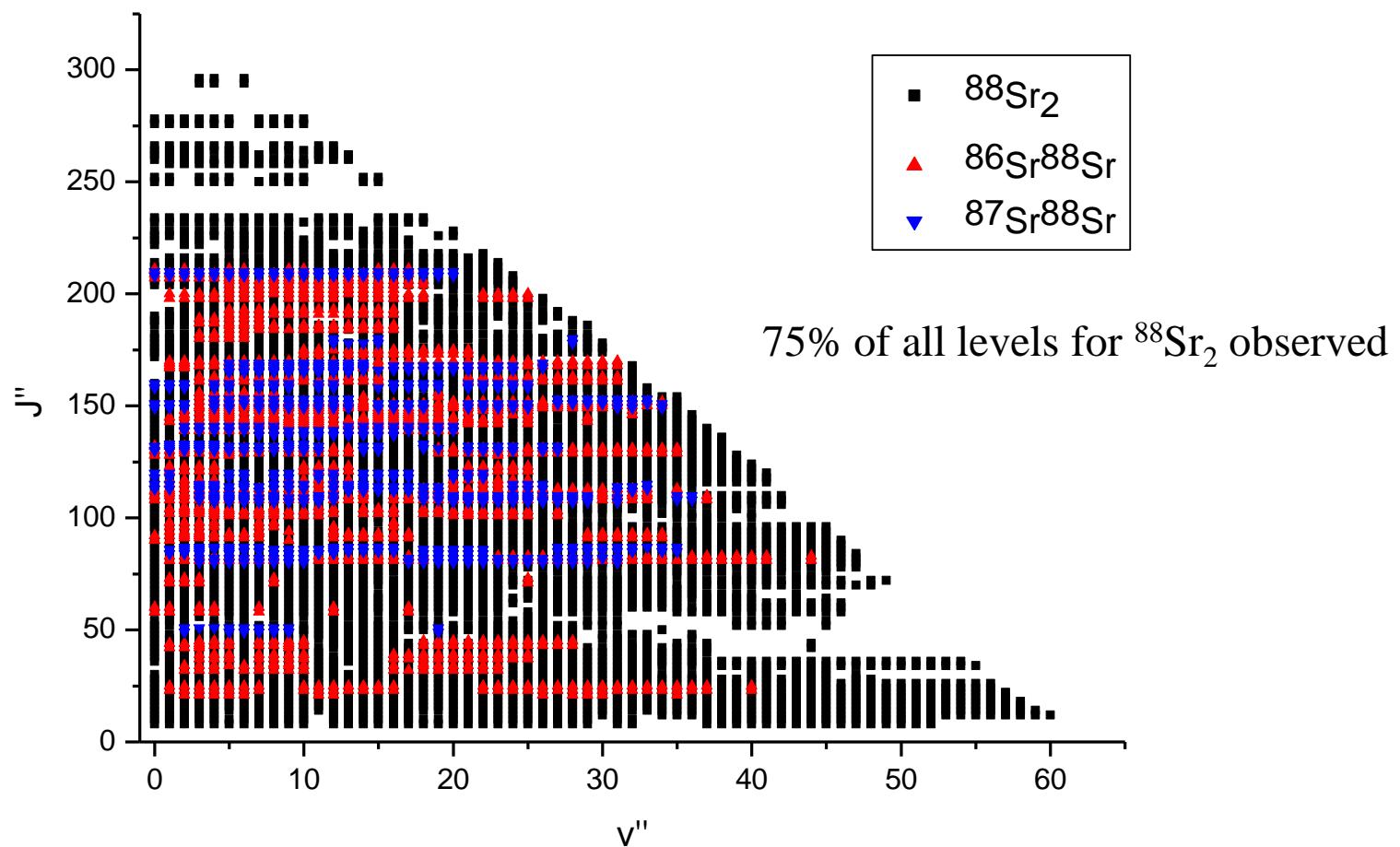


Example Sr_2 with a large data

application of molecular data → atomic clock with Sr

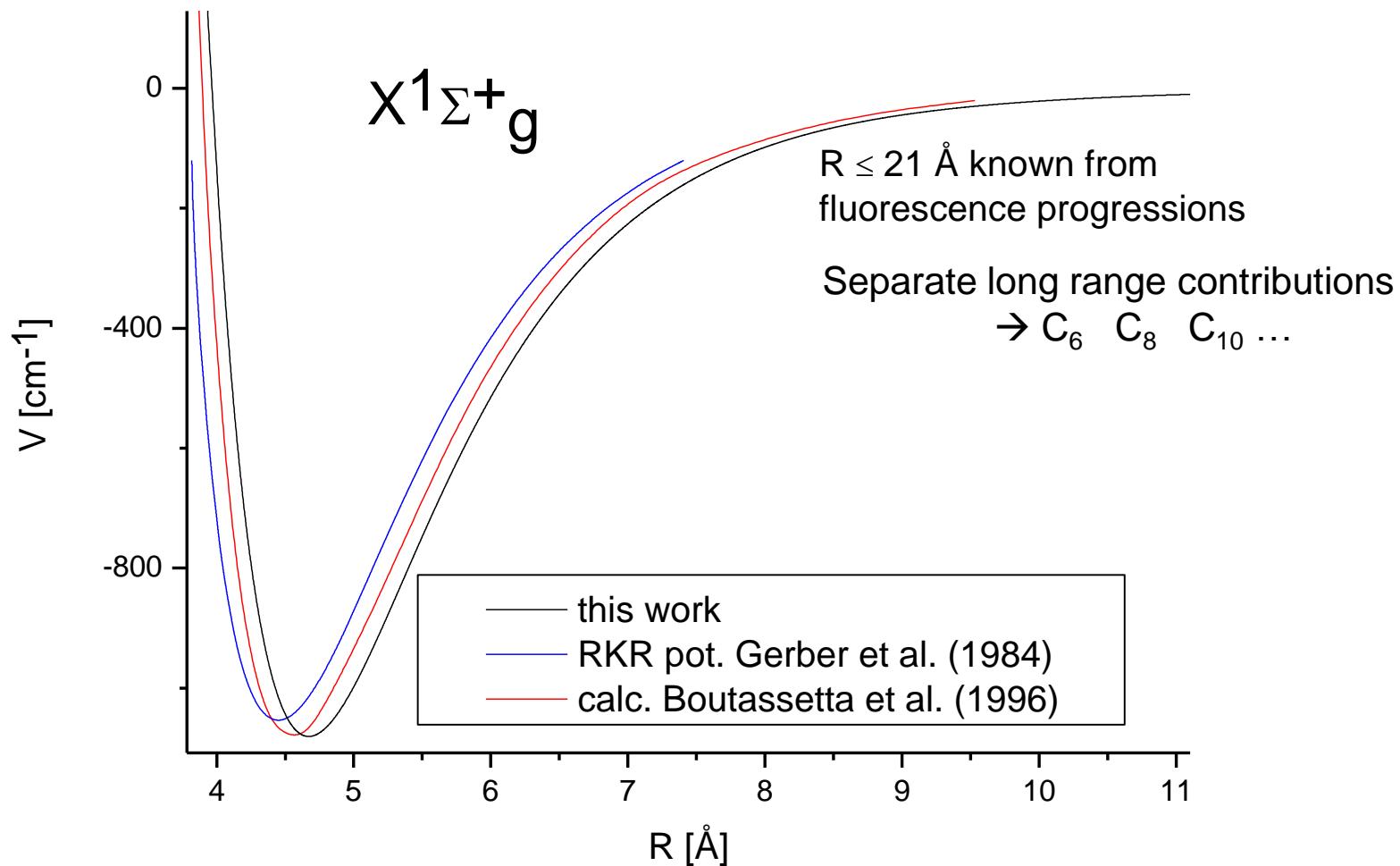
Set of molecular levels for Sr_2

area of possible levels



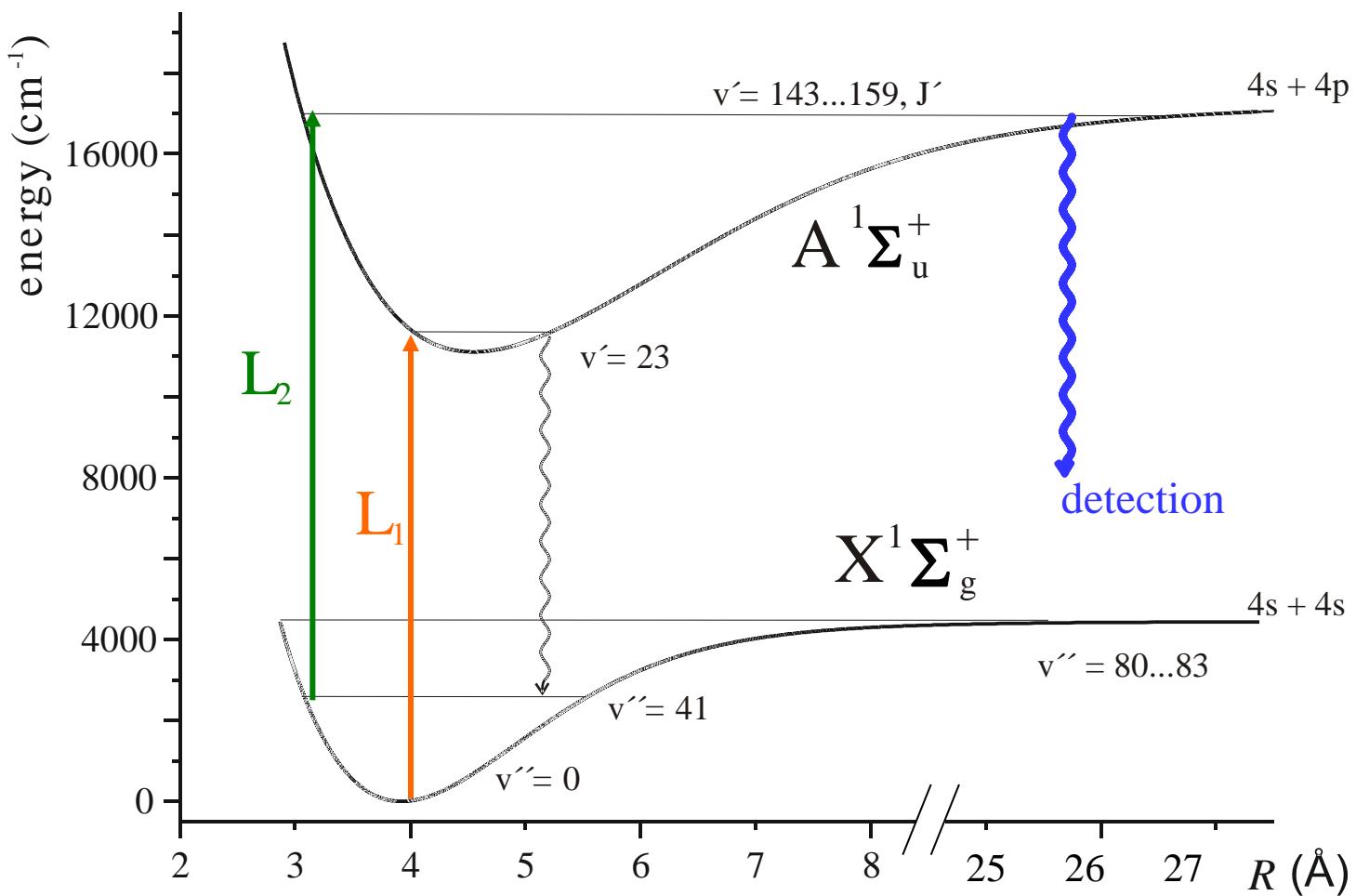
Derive a continuous function for the potential?!

Yes with 23 free parameters for 6077 data points

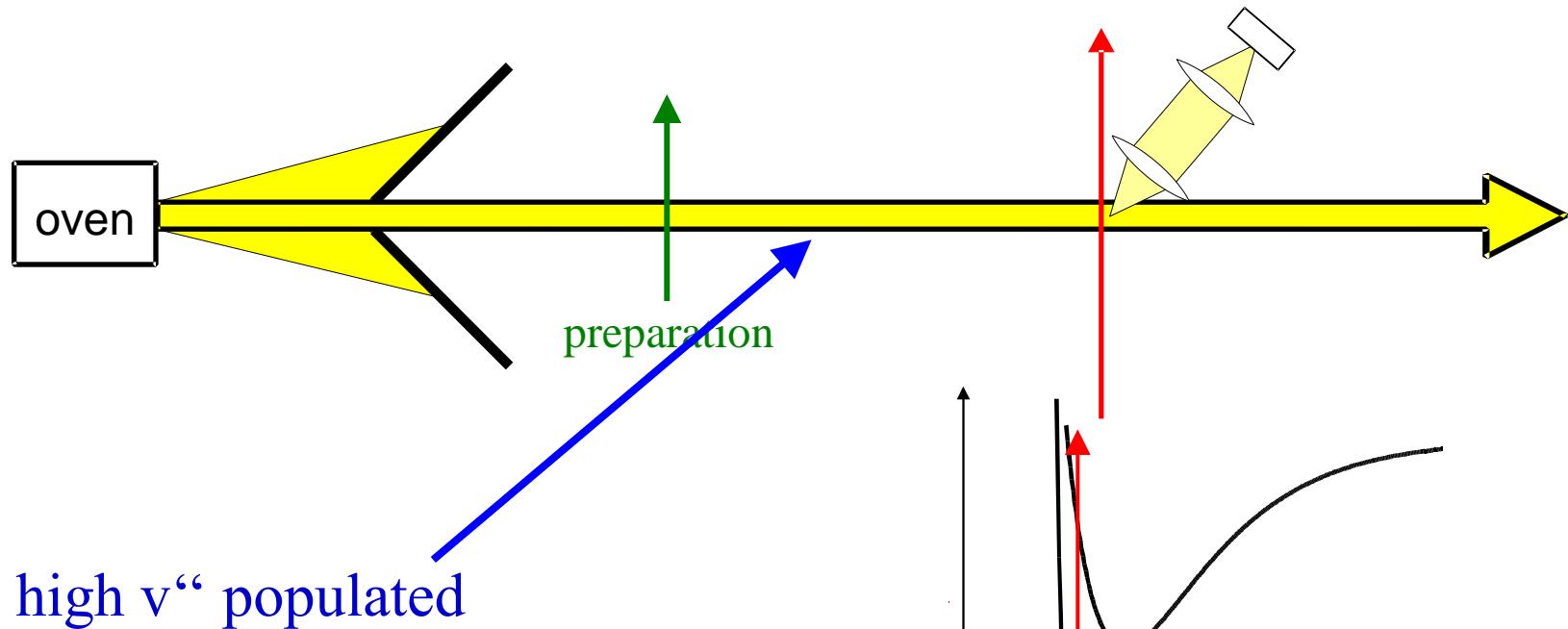


How precise are experimentally derived potentials?

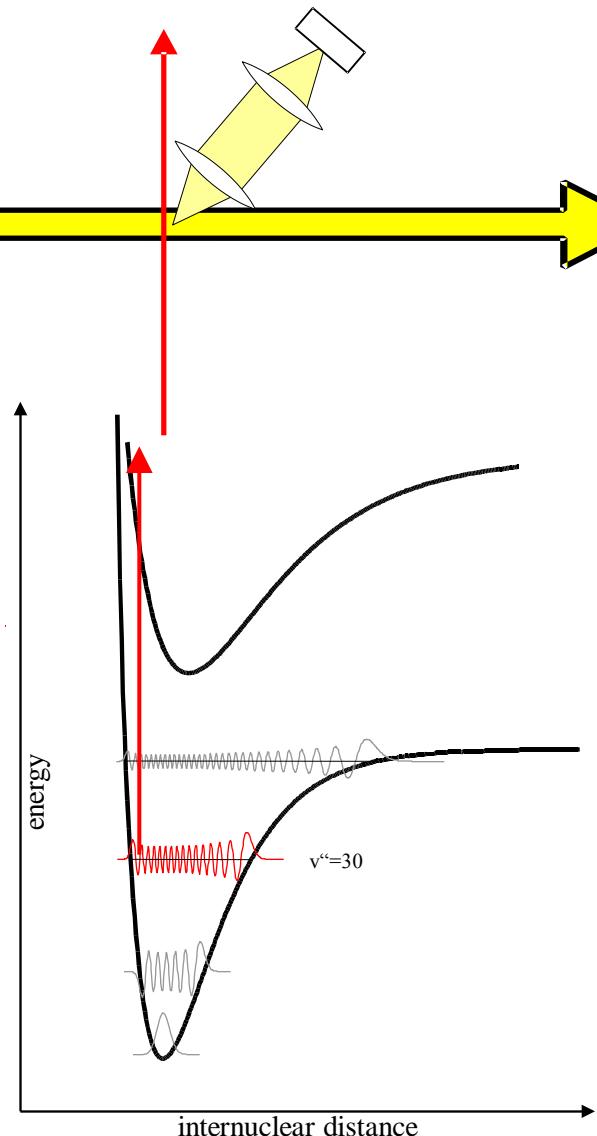
Excitation scheme for pair spectroscopy



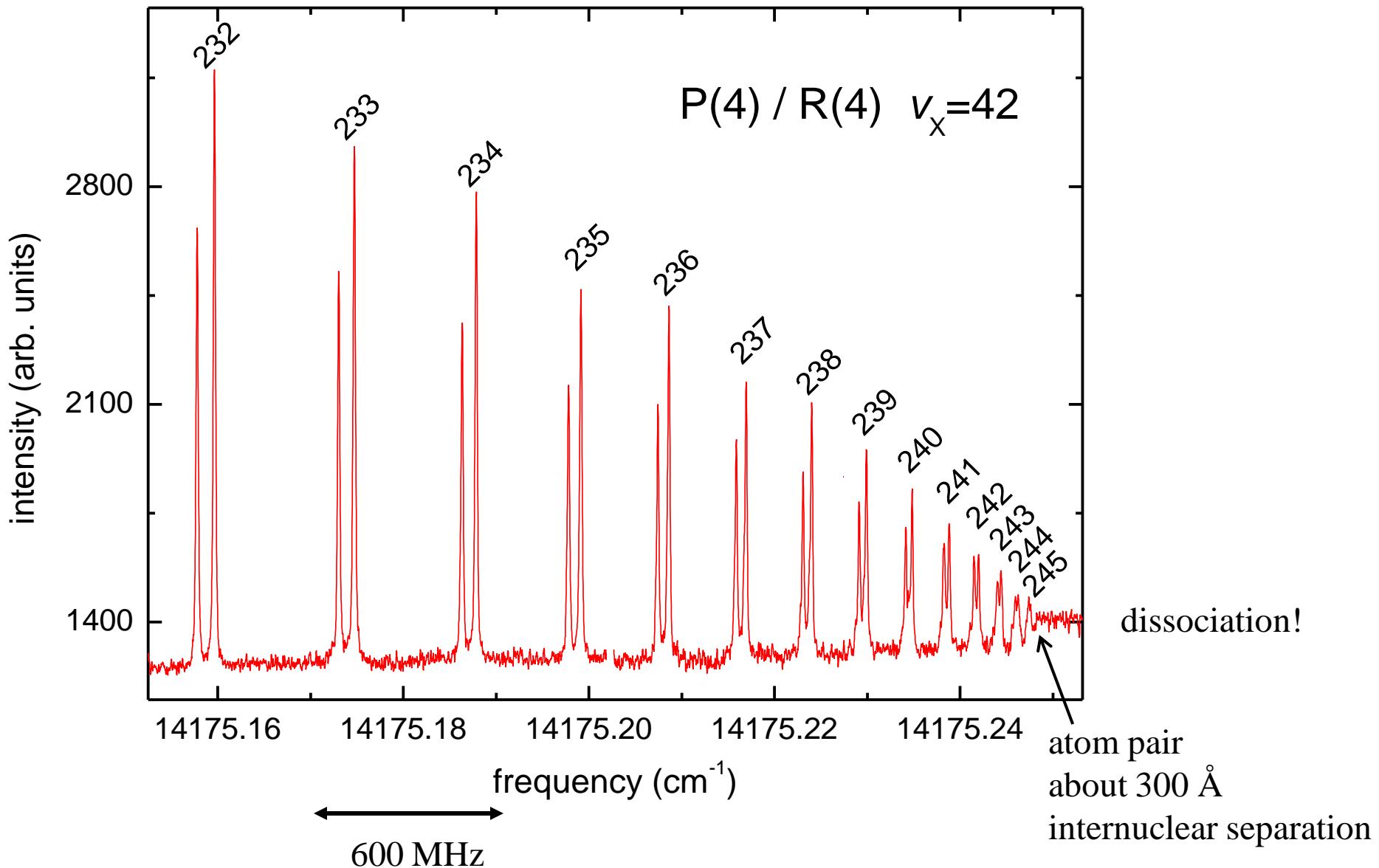
Preparing atom pairs



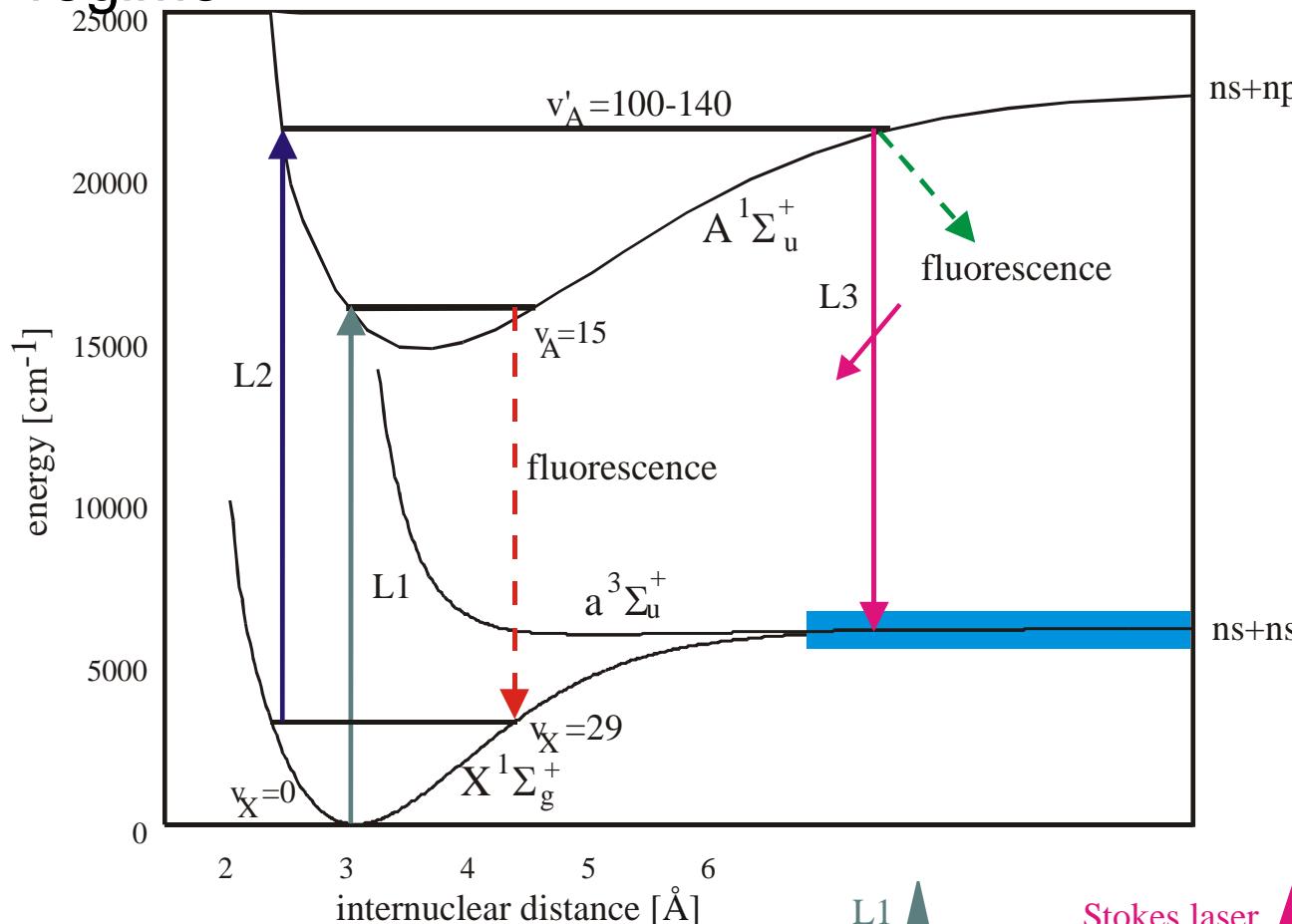
high v'' populated
asymptotic vibrational
levels become observable



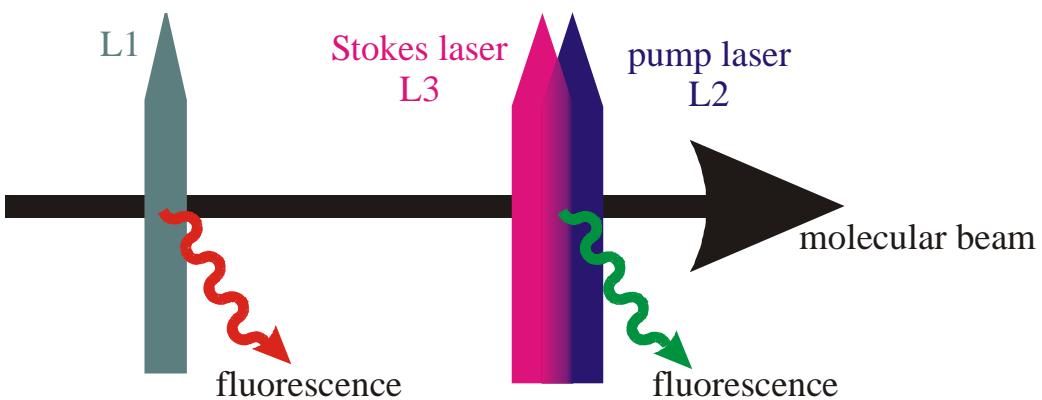
Atom pairs in K₂ beam



Coherent excitation scheme to the cold collision regime



Example Na_2

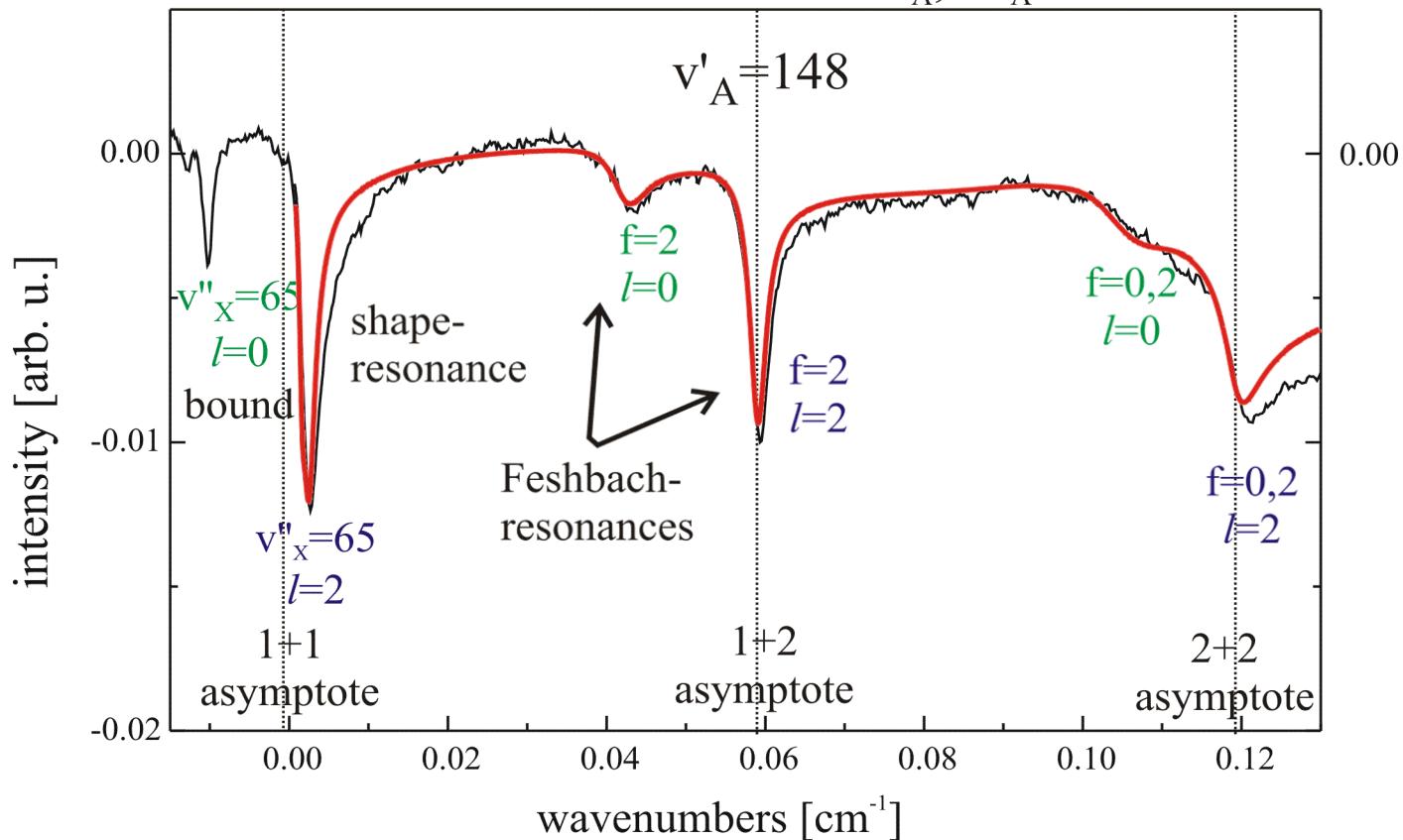


Scattering spectroscopy → inverted photoassociation



Example Na+Na

intermediate level v'_A , $J'_A=1$



Comparison with simulation

Conclusions

- studies of ultracold matter demand for new spectroscopy
 - complementary experiments in traps and with hot ensembles possible
 - evaluation of complete data from different experiments performed
-
- compact form of huge body of data → molecular potentials, Hamiltonian
 - How to keep this information for an easy application for a long time?

developing fields

- wide area of preparing quantum systems with molecules
- using the internal motion in a molecule for precise studies of its dynamic → times dependence of atomic masses?
- never complete